



PERIODIC TABLE UNIT PROJECT

You will be given a periodic table with which you will be creating a map of the elements. The purpose of this project is to understand how the periodic table is organized, what information can be obtained from the table about elements, and to become familiar with the characteristics of the various families of elements.

This is a significant project, which requires that you do some independent research to produce a quality product. You will be assessed on the completeness of your project as well as the detail and creativity with which your project is produced. **Your main goal is to produce a MAP of the elements on the periodic table that visually communicates as much information as possible to the traveler.** Be creative!! You are the cartographer (map maker)!

DUE DATE: _____

PRIMARY REQUIREMENTS:

1. Label the atomic numbers on your periodic table map.
2. Label the element names on your periodic table map.
3. Label the group numbers on your periodic table map for groups IA-VIIIA.
4. Label the period numbers on your periodic table map.
5. For the primary families, group A elements, indicate the number of valence electrons.
6. For Groups IA-VIIA, indicate the possible ionic charge.
7. Indicate which elements are found as solids, liquids, and gases at standard temperature and pressure. (*Creativity hint: what do solids, liquids, and gases look like?*)
8. Indicate which element is the most reactive element. (*Creativity hint: how could you visually represent reactivity?*)
9. Indicate the first, second, and third most abundant elements found in the Earth's crust. (*Creativity hint: how can you visually represent the crust?*)
10. Indicate at least five elements that are named after people. You must include the person's name. (*Creativity hint: what do these people look like?*)
11. Indicate at least five elements that are named after places. You must include the place name. (*Creativity hint: what do these places look like?*)
12. Indicate at least ten elements' chemical symbols that are represented by the elements' original Latin name. These symbols are ones that are unlike the element's name. (*Creativity hint: how could you represent a Latin name?*)
13. Indicate which elements have metallic properties.
14. Indicate which elements are metalloids.
15. Indicate which elements are nonmetals.
16. Indicate the alkali metals. (*On the back of the map discuss the characteristics of alkali metals*)
17. Indicate the alkaline earth metals. (*On the back of the map discuss the characteristics of alkaline earth metals*)
18. Indicate the transition metals. (*On the back of the map discuss the characteristics of transition metals*)
19. Indicate the rare earth metals. (*On the back of the map discuss the characteristics of rare earth metals and just how "rare" they are*)
20. Indicate the Lanthanoid series of elements. (*On the back of the map discuss the characteristics of these metals*)
21. Indicate the Actinoid series of elements. (*On the back of the map discuss the characteristics of these metals*)
22. Indicate the halogens. (*On the back of the map discuss the characteristics of halogens*)

23. Indicate the noble gases. (*On the back of the map discuss the characteristics of noble gases*)
24. On your map of elements, visually indicate the change in atomic size (radius) that occurs as one moves from left to right and from top to bottom along the periodic table.
25. Make sure you have provided a complete key to any colors, symbols, or labels you used. No place on the map should there be a number or symbol without an indication (in the key or elsewhere) of what that number or symbol means.
26. Works consulted list (typed and taped to the back of your map): you must cite all the resources you used in researching information for your periodic table map. You must use appropriate MLA format for all texts, websites, or computer programs used.
27. FOR EXTRA CREDIT: Incorporate any of the other periodic patterns we discussed in class (density, melting point, etc.) or add to any of the above guidelines in an extraordinary way.
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CREATIVITY HINTS:

- How can you make your periodic table look like a map?
- How can you use color, symbols, and other strategies to portray as much information as possible visually and creatively?
- How can you insert pictures of individuals or places to represent the history of the periodic table?

- Keep in mind that more does not always mean better! Strike a balance between creativity and quantity.

FORMATTING GUIDELINES:

- You can only use the blank periodic table provided. No poster board or extra sized paper allowed. You cannot include extra paper taped onto the edges in order to extend the size of your periodic table map.
- All content must be on the front of the map except for group descriptions and works consulted.
- The descriptions of the different groups of elements must be typed and taped, flush, to the back of your periodic table map. "Descriptions" are referring to physical and chemical properties. This should not be a list, rather an explanation of the characteristics of this group. You must write this in your own words and in complete sentences.
- When writing the descriptions of the element groups, you should use your own words and understand everything that you write. If you don't know what a particular characteristic is or what it means, you cannot use it in your description.
- An electronic copy of your typed descriptions must be submitted to *turnitin.com* by the time class starts on the date the project is due.
- Your periodic table map must be easily interpreted, which is why creativity is key!

EVALUATION:

- A rubric for this project will be posted online.